Amendments to the Claims:

This listing of claims will replace all prior versions, and listings, of claims in the application:

Listing of Claims:

1.-13. (Canceled)

14. (Currently Amended) A method for forming an electrode material made of substituted lithium nickel-manganese oxides, comprising the steps of: producing compositions of $\text{Li}_{1+x}\text{Ni}_{\alpha}\text{Mn}_{\beta}\text{A}_{\gamma}\text{O}_{2}$ wherein A is chosen from the group consisting of Mg, Zn, Al, Co, Ga, B, Zr, and Ti, and further wherein x is between about 0 and 0.2, α is between about 0.1 and 0.5, β between about 0.4 and 0.6, and γ between about 0 and about 0.1 through a electrode forming process chosen from the group consisting of a solid-state reaction method, an aqueous solution method, and a sol-gel method;

further wherein the solid-state reaction method comprises the steps of:

mixing amounts of lithium hydroxide (or lithium carbonate), (Ni,Mn)-hydroxide, and A-hydroxide or A oxide (wherein A is chosen from the group consisting of Mg, Zn, Al, Co, Ga, B, Zr, and Ti) in acetone using zirconia balls for about 12-24 hours to form a mixed powder;

calcining the mixed powder at about 450~550°C for about 12-30 hours in air; calcining the mixed powder at about 900-1100°C for about 10-24 hours either in an air or an nitrogen/oxygen atmosphere; and

rapidly quenching the calcined powders into liquid nitrogen.

15.-20 (Canceled)

- 21. (New) The method of claim 14, wherein x is between about 0 and 0.2, α is between about 0.2 and 0.5, and β is between about 0.5 and 0.6.
- 22. (New) The method of claim 14, wherein A is Mg and further wherein x is between about 0 and 0.2, α is between about 0.1 and 0.5, β between about 0.4 and 0.6, and γ is between about 0.01 and about 0.1.
- 23. (New) The method of claim 14, wherein A is Al and further wherein x is between about 0 and 0.2, α is between about 0.15 and 0.5, β is between about 0.45 and 0.6, and γ is between about 0.01 and about 0.1.
- 24. (New) The method of claim 14, wherein A is Co and further wherein x is between about 0 and 0.2, α is between about 0.15 and 0.5, β is between about 0.45 and 0.6, and γ is between about 0.01 and about 0.1.

- 25. (New) The method of claim 14, wherein A is Ti and further wherein x is between about 0 and 0.2, α is between about 0.2 and 0.5, the β is between about 0.4 and 0.6, and γ is between about 0.01 and about 0.1.
- 26. (New) An electronic device comprising an electrode comprised of the electrode material produced by the method of claim 14 and wherein the electrode is a cathode comprising a mixture of about 80 weight percent of the electrode material, about 10 weight percent carbon, and about 10 weight percent polyvinylidene difluoride as a binder.
- 27. (New) A method for forming an electrode material made of substituted lithium nickel-manganese oxides, comprising the steps of: producing compositions of $\text{Li}_{1+x}\text{Ni}_{\alpha}\text{Mn}_{\beta}\text{A}_{\gamma}\text{O}_{2}$ wherein A is chosen from the group consisting of Mg, Zn, Al, Co, Ga, B, Zr, and Ti, and further wherein x is between about 0 and 0.2, α is between about 0.1 and 0.5, β between about 0.4 and 0.6, and γ between about 0 and about 0.1 through a electrode forming process chosen from the group consisting of a solid-state reaction method, an aqueous solution method, and a sol-gel method;

further wherein the aqueous solution method comprises the steps of:

dissolving appropriate amounts of lithium hydroxide, nickel hydroxide, and A-hydroxide or A-nitrate (wherein A is chosen from the group consisting of Mg, Zn, Al, Co, and Ga) in water;

adding an aqueous solution of manganese acetate to form a mixed solution; refluxing the mixed solution in a round bottom flask attached with a condenser at about 80°C for about 12-24 hours;

evaporating the mixed solution in a rotary vacuum evaporator;

eliminating the organic contents in the mixed solution at about 400°C for about 2 hours;

calcining the resulting powder at about 900-1100°C for about 10-24 hours in either an air or an nitrogen/oxygen atmosphere; and

rapidly quenching the calcined powders into liquid nitrogen.

- 28. (New) The method of claim 27, wherein x is between about 0 and 0.2, α is between about 0.2 and 0.5, and β is between about 0.5 and 0.6.
- 29. (New) The method of claim 27, wherein A is Mg and further wherein x is between about 0 and 0.2, α is between about 0.1 and 0.5, β between about 0.4 and 0.6, and γ is between about 0.01 and about 0.1.
- 30. (New) The method of claim 27, wherein A is Al and further wherein x is between about 0 and 0.2, α is between about 0.15 and 0.5, β is between about 0.45 and 0.6, and γ is between about 0.01 and about 0.1.

- 31. (New) The method of claim 27, wherein A is Co and further wherein x is between about 0 and 0.2, α is between about 0.15 and 0.5, β is between about 0.45 and 0.6, and γ is between about 0.01 and about 0.1.
- 32. (New) The method of claim 27, wherein A is Ti and further wherein x is between about 0 and 0.2, α is between about 0.2 and 0.5, the β is between about 0.4 and 0.6, and γ is between about 0.01 and about 0.1.
- 33. (New) An electronic device comprising an electrode comprised of the electrode material produced by the method of claim 27 and wherein the electrode is a cathode comprising a mixture of about 80 weight percent of the electrode material, about 10 weight percent carbon, and about 10 weight percent polyvinylidene difluoride as a binder.
- 34. (New) A method for forming an electrode material made of substituted lithium nickel-manganese oxides, comprising the steps of: producing compositions of $\text{Li}_{1+x}\text{Ni}_{\alpha}\text{Mn}_{\beta}A_{\gamma}O_2$ wherein A is chosen from the group consisting of Mg, Zn, Al, Co, Ga, B, Zr, and Ti, and further wherein x is between about 0 and 0.2, α is between about 0.1 and 0.5, β between about 0.4 and 0.6, and γ between about 0 and about 0.1 through a electrode forming process chosen from the group consisting of a solid-state reaction method, an aqueous solution method, and a sol-gel method;

further wherein the sol-gel method comprises the steps of:

dissolving appropriate amounts of lithium acetate, nickel acetate, manganese acetate, A-acetate or A-nitrate; (wherein A is chosen from the group consisting of Mg, Zn, Al, Co, and Ga) in water;

adding a glycolic/tartaric acid solution for use as a chelating agent;

adjusting the pH of the resulting solution to about 7 to about 8 using ammonium hydroxide;

continuously stirring and heating the solution on a hot plate to form a gel precursor;

decomposing the gel precursor at 450°C about for about 5h in air to form a decomposed powder;

calcining the decomposed powders at about 900-1100°C for about 10-24 hours in either an air or an nitrogen/oxygen atmosphere; and

rapidly quenching the calcined powders into liquid nitrogen.

- 35. (New) The method of claim 34, wherein x is between about 0 and 0.2, α is between about 0.2 and 0.5, and β is between about 0.5 and 0.6.
- 36. (New) The method of claim 34, wherein A is Mg and further wherein x is between about 0 and 0.2, α is between about 0.1 and 0.5, β between about 0.4 and 0.6, and γ is between about 0.01 and about 0.1.

- 37. (New) The method of claim 34, wherein A is Al and further wherein x is between about 0 and 0.2, α is between about 0.15 and 0.5, β is between about 0.45 and 0.6, and γ is between about 0.01 and about 0.1.
- 38. (New) The method of claim 34, wherein A is Co and further wherein x is between about 0 and 0.2, α is between about 0.15 and 0.5, β is between about 0.45 and 0.6, and γ is between about 0.01 and about 0.1.
- 39. (New) The method of claim 34, wherein A is Ti and further wherein x is between about 0 and 0.2, α is between about 0.2 and 0.5, the β is between about 0.4 and 0.6, and γ is between about 0.01 and about 0.1.
- 40. (New) An electronic device comprising an electrode comprised of the electrode material produced by the method of claim 34 and wherein the electrode is a cathode comprising a mixture of about 80 weight percent of the electrode material, about 10 weight percent carbon, and about 10 weight percent polyvinylidene difluoride as a binder.